

Dispersing agents for electrophoretic deposition of TiO₂ and TiO₂–carbon nanotube composites

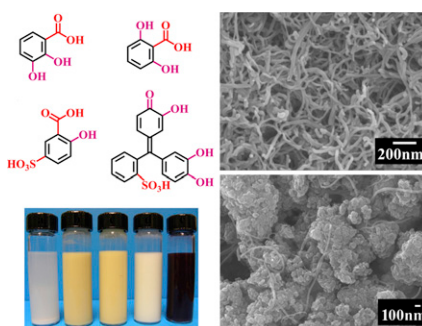
Y. Sun, Y. Wang, I. Zhitomirsky*

Department of Materials Science and Engineering, McMaster University, 1280 Main St. West Hamilton, Ontario L8S 4L7, Canada.

HIGHLIGHTS

- ▶ Molecules from catecholate and salicylate families allowed dispersion of TiO₂.
- ▶ The new anionic dispersant were used for electrophoretic deposition.
- ▶ Pyrocatecholsulfonphthalein allowed dispersion and deposition of carbon nanotubes.
- ▶ Pyrocatecholsulfonphthalein was used as universal dispersing agent.
- ▶ Composite TiO₂–carbon nanotube films were obtained by electrophoretic deposition.

GRAPHICAL ABSTRACT



ARTICLE INFO

Article history:

Received 1 September 2012
 Received in revised form
 17 November 2012
 Accepted 20 November 2012
 Available online 29 November 2012

Keywords:

Titanium oxide
 Carbon nanotube
 Electrophoretic deposition
 Dispersing agent
 Composite film

ABSTRACT

Organic anionic molecules such as 2,3-dihydroxybenzoic acid (DBA23), 2,6-dihydroxybenzoic acid (DBA26), 5-sulfosalicylic acid (SSA) and pyrocatecholsulfonphthalein (PCS) were investigated for electrophoretic deposition (EPD) of TiO₂ and TiO₂–multiwalled carbon nanotube (MWCNT) films from suspensions in ethanol. The adsorption of the molecules on TiO₂ surfaces was based on catecholate or salicylate binding, involving adjacent OH groups or adjacent OH and COOH groups, respectively. The adsorption of the anionic molecules allowed efficient dispersion, charging and EPD of TiO₂. The deposition yield was studied as a function of DBA23, DBA26, SSA and PCS concentration in TiO₂ suspensions and deposition time. An important finding was the possibility of efficient dispersion, charging and EPD of MWCNT using PCS. It was shown that PCS can be used as a universal dispersing agent for co-deposition of TiO₂ and MWCNT and fabrication of composite TiO₂–MWCNT films. The advantages of the new strategies, compared to other methods described in literature, were discussed. The deposits were studied by Fourier transform infrared spectroscopy, thermogravimetric analysis, differential thermal analysis and electron microscopy. The proposed approach paves the way for EPD of other oxide materials and composites.

© 2012 Elsevier B.V. All rights reserved.

1. Introduction

TiO₂ is an important material for electrochemical [1], photocatalytic [2,3], electronic [4,5], photovoltaic [6], biomedical [7,8] and cosmetic [9] applications. There is a growing interest in the development of TiO₂–carbon nanotube (CNT) nanocomposites with

improved electronic conductivity for solar cells [10], gas sensors [11] and catalytic devices [12]. Many applications of TiO₂ and TiO₂–CNT composites are based on the use of thin films. The choice of a film deposition method is extremely important for the fabrication of advanced materials with controlled microstructure and properties.

Electrophoretic deposition (EPD) is an attractive technique for the fabrication of oxide and composite films [13,14]. EPD is achieved via electrophoretic motion of charged particles in a suspension toward an electrode and deposit formation under the

* Corresponding author. Tel.: +1 905 5259140x23914; fax: +1 905 5289295.
 E-mail address: zhitom@mcmaster.ca (I. Zhitomirsky).