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# Design and controllable synthesis of $\alpha$ -/ $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> homojunction with synergetic effect on photocatalytic activity

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#### HIGHLIGHTS

- ► Design and synthesis of  $\alpha$ -/ $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> homojunction.
- ► The  $\alpha$ -/ $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> homojunction possess the highest photocurrent and photocatalytic activities.
- ► The characterization of the existence of  $\alpha$ -/ $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> homojunction.
- Synergic effect between  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> and  $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> in the composite.

#### ARTICLE INFO

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#### ABSTRACT

A  $\alpha$ -/ $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> composite has been successfully prepared via a hydrothermal method. The as-prepared nanocrystals are characterized by the X-ray diffraction, Fourier transformation infrared spectrum, Scanning electron microscopy, Transmission electron microscopy, and high-resolution transmission electron microscopy examination. The homojunction between the  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> and  $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> is confirmed by the diffuse reflectance spectra and impedance spectrum. The photocatalytic activities of the samples were evaluated by the degradation of Rhodamine B as a model pollutant. The composite exhibited higher photocatalytic activity than bare  $\alpha$ -Bi<sub>2</sub>O<sub>3</sub> or  $\gamma$ -Bi<sub>2</sub>O<sub>3</sub> in the degradation of RhB. The enhanced photocatalytic activity is attributed to the synergetic effect of the homojunction.

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#### 1. Introduction

Photocatalysis has been intensively investigated as it is potential in the destruction of inorganic and organic pollutants [1–3], water splitting [4,5], and carbon dioxide photoconversion [6–8]. Since visible light accounts for larger proportion of the solar spectrum, a great number of undoped single-phase oxide semiconductor photocatalysts responding to visible light have been developed, such as BiVO<sub>4</sub> [9], CaBi<sub>2</sub>O<sub>4</sub> [10], Bi<sub>2</sub>MoO<sub>6</sub> [11,12], and Bi<sub>2</sub>O<sub>3</sub> [13,14]. However, the photocatalytic activities of these photocatalysts are still not satisfying from the viewpoint of practical application. It is essential to design and synthesize more efficient visible-lightdriven photocatalysts.

Previous studies on  $TiO_2$  have shown that the photocatalytic activity of a photocatalyst is basically determined by the intrinsic properties, including crystal phase [15], defects [16], surface area [17], exposed facets [18], etc. In early 1990s, it was found the

TiO<sub>2</sub> photocatalysts with mixed phases of anatase and rutile exhibited enhanced activity compared to bare one, which was attributed to the charge transfer between anatase and rutile [19]. Many investigations have been carried out on the mixed-phase of anatase and rutile [20,21]. Xu and co-workers found that mesoporous TiO<sub>2</sub> with brookite and anatase nanocrystals exhibited higher photocatalytic activity in the degradation of acetone [22]. Li demonstrated that small rutile crystallites interweaved with anatase crystallites which benefited electron transfer at the anatase/rutile interface and thus effectively created catalytic "hot spots" [23]. Generally, a material with different phases possesses different band gap and flat band. Therefore, a homojunction could be built between different crystal phase given the two phases are in close contact, which will lead to efficient electron-hole separation and higher catalytic reactivity. Although the wide band gap of TiO<sub>2</sub> determines it to be activated under ultraviolet (UV) light only, which limited its practical application, the method of constructing a homojunction is proved to be a good way to strengthen the photocatalytic activity. These findings impulse a new beginning for the design of non-titania based visible-light driven photocatalysts with mixed phase.



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