



Removal of cadmium from aqueous solution using mesoporous PVA/TEOS/APTES composite nanofiber prepared by sol–gel/electrospinning

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HIGHLIGHTS

- ▶ The synthesized membranes were characterized by FTIR, SEM and BET analyses.
- ▶ The structure of PVA/TEOS/APTES composite nanofiber was mesoporous.
- ▶ The maximum adsorption capacity of cadmium adsorption was 327.3 mg g⁻¹.
- ▶ The mechanism of cadmium sorption onto the nanofiber is chemisorption.
- ▶ The cadmium sorption onto the composite nanofiber did not change after 8 cycles.

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ABSTRACT

The novel poly (vinyl alcohol)/tetraethylorthosilicate/aminopropyl triethoxysilane (PVA/TEOS/APTES) composite nanofiber was prepared by the sol–gel/electrospinning method and its application for the removal of cadmium from aqueous solution was investigated. The parameters of electrospinning process such as flow rate of spinning solution, applied voltage and tip–collector distance (TCD) were optimized for fabrication of homogeneous and fine fibers. As results, the fibers with the smallest average diameter of about 196.70 nm were obtained at the flow rate of 0.3 mL h⁻¹, applied voltage of 17.5 kV and TCD of 12.5 cm. Fourier Transform Infrared (FTIR) revealed the presence of amine groups in the structure of PVA/TEOS/APTES composite nanofiber. The result of pore volume calculation by BJH model showed that the structure of PVA/TEOS/APTES nanofiber was mesoporous. Experimental parameters affecting cadmium sorption onto the PVA/TEOS/APTES nanofiber such as pH, contact time, initial concentration and temperature were studied. The kinetic data were analyzed by the pseudo-first-order, pseudo-second-order and double-exponential kinetic models. The Langmuir, Freundlich and Redlich–Peterson isotherm models were applied to describe the equilibrium data of cadmium sorption onto the composite nanofiber. Thermodynamic parameters indicated that the cadmium adsorption onto the composite nanofiber was endothermic and spontaneous process. The reusability of the composite nanofiber was also determined after eight sorption–desorption cycles.

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1. Introduction

Sol–gel technique is a suitable method for the preparation of organic–inorganic hybrids from precursors containing alkoxysilyl groups through steps of hydrolysis and condensation [1]. The silica-based adsorbents such as TEOS due to the large specific surface area were used widely in the removal of heavy metal ions [2,3]. Furthermore, the removal efficiency of metal ions increases

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remarkably in the wake of the adsorbent surface modification by the functional groups such as –NH₂, –SH and –S– groups [2,3]. On the other hand, electrospinning technique is an efficient method for fabrication of sub-micron or nanoscale fibers [4–7]. Nanoscale fibers of silica materials prepared by sol–gel/electrospinning technique have a large surface area and high porosity which these properties provide high separation efficiency. In recent years, researchers used from polyvinylpyrrolidone (PVP)/SiO₂ [8], PVA/SiO₂ nanofiber membrane [9] and amidoxime-modified polyacrylonitrile (PAN-oxime) [10] for adsorption of heavy metals from aqueous solutions.

In electrospinning technique, high voltage is applied between a nozzle and a collector where an electrically charged jet of polymer