



Graphene facilitated visible light photodegradation of methylene blue over titanium dioxide photocatalysts

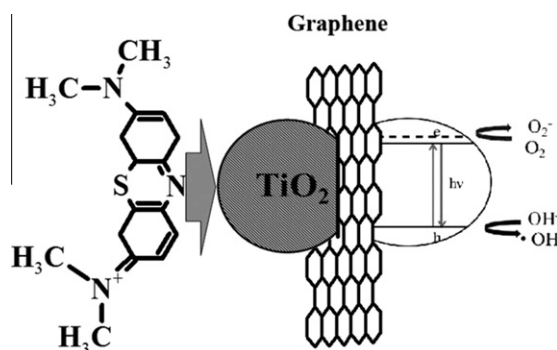
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HIGHLIGHTS

- ▶ Graphene–titania composites (G–TiO₂) were synthesized by a sol–gel method.
- ▶ G–TiO₂ catalysts showed high efficiency in degradation of methylene blue under visible light.
- ▶ *In situ* prepared G–TiO₂ presented higher activity than that of G–TiO₂(P25).
- ▶ The mechanism of graphene in the enhanced visible light photocatalytic activity was proposed.

GRAPHICAL ABSTRACT



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ABSTRACT

Several graphene–titania composites (G–TiO₂) were synthesized by a sol–gel method using titanium isopropoxide (or P25) as Ti-precursors and reduced graphene oxide (RGO). The structural, morphological, and physicochemical properties of the samples were thoroughly investigated by X-ray diffraction (XRD), Fourier transform infrared spectroscopy (FT-IR), field emission scanning electron microscopy (FE-SEM), UV–vis diffuse reflectance (UV–vis DRS), and thermogravimetric-differential thermal analysis (TG-DTA). A significant increase in light absorption to visible light was observed by G–TiO₂ compared with that of naked TiO₂. The photocatalytic activity of G–TiO₂ in methylene blue bleaching under visible light (>430 nm) is much enhanced. G–TiO₂ synthesized from titanium isopropoxide hydrolysis presented higher activity than that of G–TiO₂(P25). Contribution of graphene on the enhancement of visible-light photocatalytic activity of the composite was discussed.

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1. Introduction

Photocatalytic decomposition of various organic compounds in aqueous solutions has been widely studied and many nanomaterials have been developed as photocatalysts for this technology [1–4]. TiO₂ has been intensively investigated as a photocatalyst for environmental clean-up and solar energy conversion. However, TiO₂ can only decompose aromatic organics into CO₂ and H₂O under UV-illumination and suffers from a barrier in responding

to visible light at wavelengths higher than 387 nm due to a large band gap of 3.2 eV. As a result, only 3–5% of the solar energy that reaches onto the earth surface can be utilized. The common strategies for extending the absorption threshold of TiO₂ to visible light region include doping, coupling or anchoring with other organic or inorganic elements such as nitrogen, carbon, halogen, and metals into the titania lattice [5–11].

Combination of different types of carbon with TiO₂ has been suggested as a promising method for an enhanced photocatalytic performance [12,13]. In the past a few years, graphene as a novel carbonaceous nanomaterial has attracted more and more interests due to its unique and excellent performance in chemical, structural

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