



Catalytic growth of high quality single-walled carbon nanotubes over a Fe/MgO catalyst derived from a precursor containing Feitknecht compound

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ABSTRACT

A catalyst is derived from a Feitknecht-compound (FC) precursor prepared via a co-precipitation method and is proved to manifest a strong metal–support interaction which facilitates the growth of high quality single-walled carbon nanotubes (SWCNTs) with a narrow diameter distribution. The effect of the rinsing condition of the precipitate on the growth of the SWCNTs is also examined. The pore structure of the catalysts rinsed by ethanol is modified and the carbon yield is enhanced while the diameter distribution of the SWCNTs is unchanged. The correlation between the metal–support interaction and the SWCNT growth is discussed.

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1. Introduction

Single-walled carbon nanotubes (SWCNTs) have aroused a great research interest due to their unique mechanical, electrical and thermal conductive properties [1,2]. The traditional SWCNT production methods include carbon-arc discharge [3], laser ablation [4] and chemical vapor deposition (CVD) [5–7]. CVD can be carried out continuously at a relatively low temperature and has been thought to be the most promising process for large-scale production [8]. However, mass production of the high quality SWCNTs is still a challenge which needs a rational catalyst design and reactor optimization.

Supported metal catalysts with Fe, Co and Ni as the active components have been widely used for the growth of the carbon nanotubes (CNTs) and carbon nanofibers [7,9–12]. Metal–support interaction (MSI) in the supported metal catalysts plays an important role in CNT growth. Generally, a strong MSI decreases the mobility of the metal particles on the support surface and enhances the metal dispersion [13]. The size of the metal nanoparticle has a direct effect on the number of walls and the diameter distribution of the CNTs. When the size of the metal nanoparticle is smaller than 5 nm, the main products are single and double-walled CNTs (DWCNTs), while larger metal nanoparticles lead to multi-walled CNTs (MWCNTs) [14]. Mattevi et al. [15] compared Fe

catalysts supported on Al₂O₃ and SiO₂ and noted that a stronger MSI between Fe and Al₂O₃ restricts the Fe nanoparticle mobility and results in a much smaller Fe particle size. Ago et al. [16] found that after a thermal treatment in Ar, the iron nanoparticles become smaller and are appropriate for the SWCNT growth. They believed that Fe atoms diffuse into the MgO lattice during the heat treatment which results in a strong MSI. Ning et al. [17] found that a uniform MgFe₂O₄/MgO solid solution formed in the Fe/MgO catalyst after 900–1000 °C calcination also manifests a strong MSI. However, this high temperature treatment leads to a low surface area and also a low SWCNT yield. An ideal catalyst for SWCNT production should have a large specific surface area with a strong MSI.

Feitknecht compound (FC) has been recognized as a good precursor for many catalysts since 1970s [18–20]. The catalyst prepared via FC structured precursor enables a strong MSI in the reduced catalyst, and facilitates the formation of the mixed oxide with a high uniformity and large total metal surface area. Ni–Cu–Al catalysts derived from FC structure have been applied in the carbon nanofiber growth via methane catalytic decomposition reaction [21–26]. And also, Fe–Mg–Al, Co–Mg–Al and Ni–Mg–Al catalysts derived from FC structures were also employed as the catalysts for the SWCNT growth in a fluidized bed reactor [27]. However, the removal of Al₂O₃ support material from produced SWCNTs during the purification process is a laborious work.

Magnesium oxide supported iron catalyst (Fe/MgO) has been used as the catalyst for SWCNT production [13,16,17,28]. Fe/MgO catalyst is environmentally benign and attractive in the economic aspects. And also, Fe and MgO can be easily removed by acid,

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