



Synthesis and photocatalytic properties of nanocrystalline Au, Pd and Pt photodeposited onto mesoporous RuO₂-TiO₂ nanocomposites

Adel A. Ismail^{a,c,*}, Detlef W. Bahnemann^b, Saleh A. Al-Sayari^c

^a Advanced Materials Department, Central Metallurgical R&D Institute, CMRDI, P.O. Box: 87, Helwan 11421, Egypt

^b Institut für Technische Chemie, Leibniz Universität Hannover, Callinstrasse 3, 30167 Hannover, Germany

^c Centre for Advanced Materials and Nanoengineering (CAMNE), Najran University, P.O. Box: 1988, Najran 11001, Saudi Arabia

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ABSTRACT

Noble metals (Au, Pd and Pt) have been photodeposited onto hexagonal mesoporous RuO₂-TiO₂ nanocomposites to study their influence on the photocatalytic activity under UV and visible lights by determination of the formation rate of HCHO generated by photooxidation of CH₃OH in aqueous solution. X-ray diffraction (XRD) patterns and N₂ sorption isotherms reveal that highly crystalline TiO₂ and mesoporous structure have been formed with high surface area (150–180 m²/g) and pore diameter ranging from 6.8 to 7.3 nm. TEM measurements show that the framework of the highly crystalline mesoporous RuO₂-TiO₂ is composed of anatase phase grown along [1 0 1] direction. The dependence of HCHO formation rate on the noble metals/RuO₂-TiO₂ nanocomposites, behaves quite differently depending on noble metals as electrons sink. Under UV light, the findings reveal that Pd/RuO₂-TiO₂ offers an improvement in term photooxidation rate of CH₃OH and photonic efficiencies over Pt/RuO₂-TiO₂ and Au/RuO₂-TiO₂. However, under visible light, the photocatalytic activity of mesoporous RuO₂-TiO₂ containing Au nanoparticles towards CH₃OH oxidation is remarkable and the photonic efficiency of RuO₂-TiO₂ has been improved two times. Pt and Pd nanoparticles could not be observed in any improvement in photocatalytic activity of RuO₂-TiO₂ under visible light.

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1. Introduction

TiO₂ anatase can only be excited by UV irradiation ($\lambda < 380$ nm) because of its large band gap energy of 3.2 eV. Moreover, the rapid recombination of photoinduced electrons and holes greatly lowers the quantum efficiency [1]. Therefore, it is of great interest to improve the generation and separation of photoinduced electron-hole pairs in TiO₂ for further applications. The manipulation of semiconductor heterostructures is one of the effective methods for photoinduced electron-hole generation and separation in recent years [2,3]. Multiple-semiconductor devices can absorb a larger fraction of the solar spectrum, which is beneficial for the excitation of the semiconductor and thus the photoinduced generation of electrons and holes. Moreover, the coupling of two different semiconductors could transfer electrons from an excited small band gap semiconductor into another attached one in the case of proper conduction band potentials [4–6]. This favors the separation of photoinduced electrons and holes and thus improves

the photocatalytic efficiency of semiconductor heterostructure dramatically.

The metal-semiconductor (MS) contact is one of the most widely used rectifying contacts in the electronics and photocatalysis applications [7]. When a metal and a semiconductor are joined to form MS interface, a significant redistribution of charge is expected to take place due to overlap of wave functions from two sides [7b]. Due to the technological importance of Schottky barrier and the most simple of the MS contact devices which are of the importance in the electron-hole separation, and, indeed, in the photocatalytic process. The nonideal behavior observed in Schottky barrier diodes has been generally attributed to the effect of interface states and the interfacial layer, which are present between metal contact and semiconductor. Thereby, the performance and reliability of Schottky barrier diodes generally depend on interface state density and their energy distribution [8]. Doping of noble metals (Au, Pd, Ag, Pt) with mesoporous TiO₂ photocatalysts was proposed to enhance the photocatalytic activity due to their different Fermi levels, characterized by the work function of the metals and the band structure of the semiconductors. Upon contact, a Schottky barrier can be formed between the TiO₂ and the noble metals, leading to a rectified charge carrier transfer [9–12].

3D mesoporous TiO₂ network acts as an antenna system transferring the initially generated electrons from the location of light

* Corresponding author at: Advanced Materials Department, Central Metallurgical R&D Institute, CMRDI, P.O. Box 87, Helwan 11421, Egypt. Tel.: +20 225010643; fax: +20 225010639.

E-mail address: aismail@cmrdi.sci.eg (A.A. Ismail).