



Short Communication

Influence of nano-ZrO₂ additive on the bending strength and fracture toughness of fluoro-silicic mica glass–ceramicsHong Yang^a, Shun Wu^a, Jiang Hu^b, Zhongyi Wang^a, Ran Wang^c, Huiming He^{a,*}^a Department of Prosthodontics, School of Stomatology, Fourth Military Medical University, Xi'an 710032, PR China^b Department of Stomatology, Affiliated Hospital, Jilin Medical College, Jilin 132013, PR China^c Fengtai Clinic of Official Outpatient Department of Navy, Beijing 100071, PR China

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ABSTRACT

Fluoro-silicic mica glass–ceramics were prepared by a sintering process and different proportions of nano-ZrO₂ particles (3Y-TZP) were integrated during the process. Bending strength and fracture toughness were evaluated using a three-point bending test and a Vickers indenter, respectively. The bending strength and fracture toughness improved significantly with the increase in the quantity of nano-ZrO₂ additives. The highest bending strength of 324.3 ± 12.3 MPa and fracture toughness of 4.2 ± 0.11 MPa m^{1/2} were obtained with 30% (wt.) nano-ZrO₂. Good results were also obtained in morphological observations. The glass–ceramic is homogenous and the ZrO₂ grains embed in the lamellar structures of the fluoro-silicic mica homogeneously and completely and array well and compactly. On the fracture surface, both the transgranular fracture and the intergranular fracture can be observed clearly.

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1. Introduction

Dental ceramic materials possess good biological properties and natural tooth-like color and luster, and thus have potential for providing excellent aesthetic qualities for dental prostheses. Fluoro-silicic mica glass–ceramic is a common dental glass–ceramic material in which lamellar mica crystals are evenly spread in a glass substrate. The phase interfaces in this compound are relatively weak and the crystal particles are dissociated easily, making the material machinable [1,2]. Fluoro-silicic mica glass–ceramics are thus commonly applied in advanced dental computer aided design/computer aided manufacture (CAD/CAM) systems for processing dental prostheses [3–5]. However, high fragility and low strength make it difficult for fluoro-silicic mica glass–ceramics to meet the demands of the mechanical properties (bending strength: 300 MPa) required for dental prostheses [6,7]. These disadvantages have hampered its application in the dental clinical field, and currently it is only used to process veneers, in-lays and crowns with a low bending strength requirement.

Studies have shown that Zirconium dioxide (ZrO₂) can toughen glass–ceramics. ZrO₂ has been applied in the preparation of baria-silica system glass–ceramics, lithium–silica system glass–ceramics and fluoro-calcium mica glass–ceramics [8,9]. However, there are

no reports on the application of ZrO₂ in fluoro-silicic mica glass–ceramics.

Melting and sintering are the main processes used in the preparation of fluoro-silicic mica glass–ceramic blocks used in the dental CAD/CAM systems. Compared with the melting process, the sintering process has the advantage of having a lower processing temperature, shorter processing time and a higher proportion of crystal phases. It is also easier to integrate other useful materials into the glass–ceramic during the sintering process. Fluoro-silicic mica glass–ceramics (KMg_{2.5}Si₄O₁₀F₂) were prepared here using the sintering method, and the toughening effect of nano-ZrO₂ particles was evaluated.

2. Experimental

2.1. Sample preparation

Analytical reagent powders of raw materials were mixed in the proportion [13.5 MgF₂–2.25 Al₂O₃–4.5 B₂O₃–3 P₂O₅–15 MgO–84 SiO₂–21 K₂O–4.5 rest (wt.%)]. The mixed powder was melted in a platinum crucible at 1475 °C for 2 h and then the melt was quenched in cold water to obtain the glass frit. Nanoscale ZrO₂ particles (Nano-ZrO₂, particle size: 50 nm) of Y₂O₃ (3.0 mol.%) stabilized tetragonal Zirconia polycrystalline (3Y-TZP) were added to the glass frit in the different proportions, which were 0 wt.% (ZrO₂-0 group), 5 wt.% (ZrO₂-5 group), 10 wt.% (ZrO₂-10 group), 20 wt.% (ZrO₂-20 group) and 30 wt.% (ZrO₂-30 group) as shown in Table 1.

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