



Modeling and Optimization of green synthesis of nickel nanoparticles via Phlomis cancellata Bunge extract through Response Surface Method

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Abstract

Green synthesis of metal nanoparticles is an interesting issue of nanoscience due to its simplicity and eco-friendliness. The present study describes a cheaper, non-toxic and simple route for biosynthesis of nickel nanoparticles using *Phlomis cancellata* Bunge extracts. Since the experimental conditions of this procedure play vital roles in the synthesis rate of the nanoparticles, a response surface methodology using the central composite design was employed for testing the reaction variables. The individual and interactive effects of process variables (temperature, time, concentration of $\text{Ni}(\text{NO}_3)_2$ and pH) upon extracellular biological synthesis of NiNPs by *Phlomis cancellata* Bunge were studied. The statistical and perturbation plot analysis suggest that a reaction temperature of 40°C , duration of 30 min., pH of 9.0 and concentration of 26 mM of $\text{Ni}(\text{NO}_3)_2$ would produce the highest amount of nanoparticles. The nanoparticles were characterized by UV-Visible spectrophotometry.

Keywords: response surface methodology, green synthesis, nickel nanoparticles, *Phlomis cancellata* Bunge.