

REVIEW

Conducting polymer-silver composites

Jaroslav Stejskal*

Institute of Macromolecular Chemistry, Academy of Sciences of the Czech Republic, 162 06 Prague 6, Czech Republic

Received 10 July 2012; Revised 22 October 2012; Accepted 25 October 2012

Preparations of hybrid composites composed of two conducting components, a conducting polymer and silver, are reviewed. They are produced mainly by the oxidation of aniline or pyrrole with silver ions. In another approach, polyaniline or polypyrrole are used for the reduction of silver ions to metallic silver. Other synthetic approaches are also reviewed. Products of oxidation of aniline derivatives, including phenylenediamines, are considered. Morphology of both the conducting polymers and the silver in composites displays a rich variety. Conductivity of the composites seldom exceeds 1000 S cm⁻¹ and seems to be controlled by percolation. Interfacial effects are also discussed. Potential applications of hybrid composites are outlined; they are likely to extend especially to conducting inks, printed electronics, noble-metal recovery, antimicrobial materials, catalysts, and sensors.

© 2012 Institute of Chemistry, Slovak Academy of Sciences

Keywords: polyaniline, polypyrrole, poly(*o*-phenylenediamine), poly(*p*-phenylenediamine), silver, silver nanoparticles, hybrid composites, conductivity

Introduction	815
Polyaniline–silver composites	816
Oxidation of aniline with silver	
compounds	816
Physical acceleration	817
Chemical acceleration	818
Morphology of composites	818
Reduction of silver ions with PANI	820
Effect of counter-ions	821
Silver complexes	821
Morphology of PANI	821
PANI films and coatings	821
Preparation of PANI in the presence of	
silver particles	822
Reduction of silver ions with external	
reductants in the presence of PANI	823
Mixing of PANI and silver particles	823
More complex systems	823
Two reductants of silver ions	823
Two oxidants of aniline	824
Ternary composites	824
Colloids	824
Polypyrrole–silver composites	824

Oxidation of pyrrole with silver	
compounds	824
Morphology of composites	825
Reduction of silver ions with PPy	826
Preparation of PPy in the presence	
of silver particles	827
Reduction of silver ions with external	
reductants in the presence of PPy	827
Mixing of PPy and silver particles	827
More complex systems	827
Multiple reactants	827
Ternary composites	827
Colloids	828
Related polymer–silver composites	828
Poly(p-phenylenediamine)	829
Poly(m-phenylenediamine)	829
Poly(o-phenylenediamine)	829
Substituted polyanilines	830
Polymethylanilines	830
Poly(o-methoxyaniline)	830
Poly(2,5-dimethoxyaniline)	830
Poly (4-aminodiphenylamine)	830
Sulfonated polyaniline	830

 $\label{eq:corresponding} \ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\xspace{^{\circ}}\ensuremath{^{\circ}}\$