International Journal of New Chemistry, 2017, 4 (4), 111-117 Published online January 2018 in http://www.ijnc.ir/. Original Article



Investigation of high-energy heterocyclic synthesis as a green fuel from the reaction of 3,6-D-aminotrazine with nitric acid and sodium azid; under different temperature conditions, by DFT method

Fatima Alzahar*

Department of Chemistry, Bishop Heber College (Autonomous), Tiruchirappalli-620 017, Tamil Nadu, India. *Corresponding Author e-mail Address: F.alzahar.chem@yahoo.com
Received 1 September 2017; Accepted 27 October 2017; Published 1 December 2017

Abstract

In this research, the synthesis of heterocyclic explosives (ATTz) from the reaction of 3,6-D-amino-tetrazine with Nitric acid and sodium azide were studied under different temperature conditions using the functional density theory method. For this purpose, the materials were first geometric optimization reaction sides, then the thermodynamic parameters were calculated for all of them. Then, the values of ΔH , ΔG , ΔS of this reaction were obtained at different temperatures as the sum of these parameters in the products to the raw materials. Finally, the best temperature for the synthesis of explosives was evaluated according to the thermodynamic parameters.

Keywords: Explosive, ATTz, Synthesis, 3,6-D-Amino tetrazine.

1. Introduction