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Original Research Article

Computational Review of Conformers 2- Chloroacetaldehyd

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ABSTRACT

Abstract In studies performed on 2- chloroacetaldehyde by abinitio beginning with the LC-WPBE, B3LYP and m06-2x functionals and 6-311++G (d, p) basis set. Also, by checking the total energy, HOMO-LUMO gaps and dipole moment, it was found that the keto form is more stable than the enol form. Theoretical calculations With LC-WPBE/6-311G(d,p) shows the more stability of I- conformer in compared to other conformers . NBO analysis was practical for illustrating the negative hyperconjugative effect on the conformers. In the basis of NBO analysis, the $LP(2)O \rightarrow \sigma^*(C-H)$ and $LP(2)O \rightarrow \sigma^*(C-C)$ Interactions were responsible of the negative hyperconjugation in the examined compounds. The deletion of all thw donor – acceptor electronic interaction from the fock matrices and off –diagonal elements, values of these interaction were reported.

Keywords: 2- chloroacetaldehyd, NBO (natural bond orbital analysis), Negative hyperconjugative effect, Second order the stabilization energies (E_2)